

# Exposure Dose from Natural and Artificial Radiation

## Natural radiation (in Japan)

From outer  
space  
**0.3mSv**



From foods  
**0.99mSv**

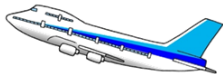


From radon and  
thoron in the air  
**0.47mSv**

From the ground  
**0.33mSv**

Annual dose from natural radiation (Japanese average): 2.1 mSv

Annual dose from natural radiation (global average): 2.4 mSv



Tokyo to New York  
Air travel (round trip) **0.08~  
0.11mSv**

## Artificial radiation



CT scan  
(single scan) **2.4~  
12.9mSv**

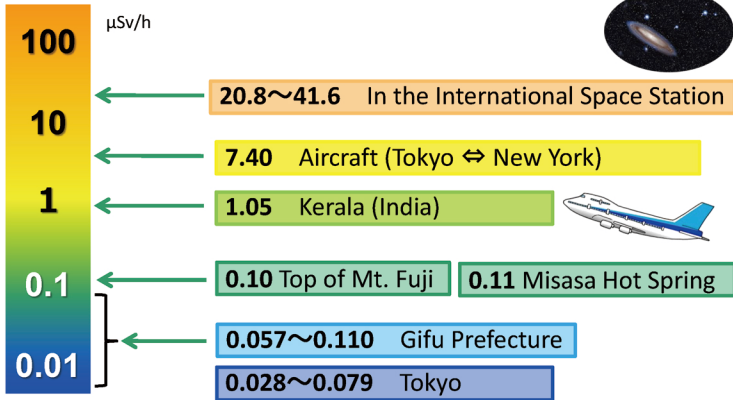


Chest X-ray scan (single scan) **0.06mSv**

mSv: millisieverts

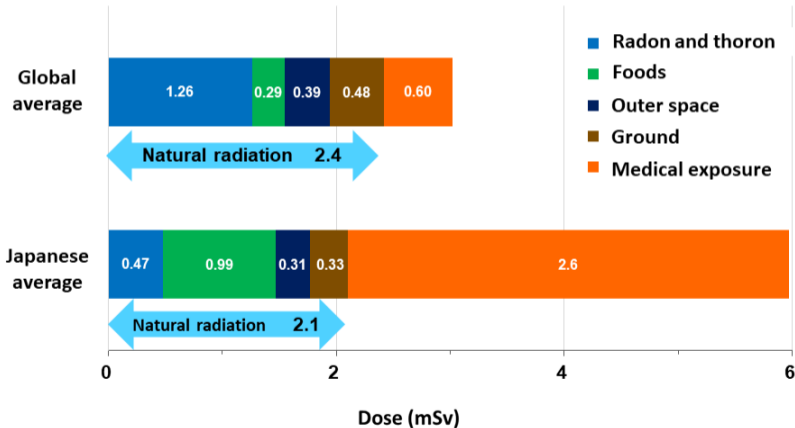
Sources: Prepared based on the 2008 UNSCEAR (United Nations Scientific Committee on the Effects of Atomic Radiation) Report; "Environmental Radiation in Daily Life (Calculation of National Doses), ver. 3" (2020), Nuclear Safety Research Association; and ICRP (International Commission on Radiological Protection) 103, etc.

## Comparison of ambient dose rates



Sources: Prepared based on "Radiation Exposure Management," the website of the JAXA Space Station Kibo PR Center, 2013; "Japanese Internet System for Calculation of Aviation Route Doses (JISCARD)," the website of the National Institute of Radiological Sciences; "Research on Ambient Gamma-ray Doses in the Environment," the website of the National Institute of Radiological Sciences; Furuno, p.25-33 of the 51st report of the Balneological Laboratory, Okayama University, 1981; and Nuclear Regulation Authority Radiation Monitoring Information (range of previous average values at monitoring posts)

## Exposure in daily life (annual)



# Breakdown of Natural Exposure Doses (Japanese)

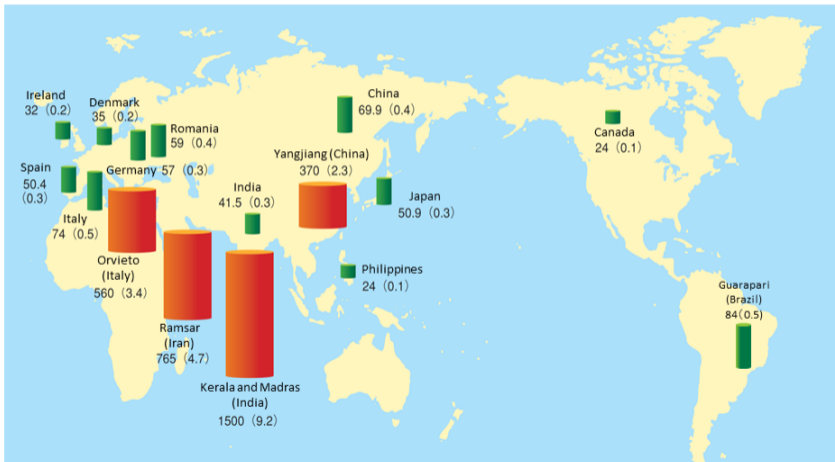
Type of exposure	Breakdown of radiation sources	Effective dose (mSv/year)
External exposure	Cosmic rays	0.3
	Ground radiation	0.33
Internal exposure (inhalation)	Radon-222 (indoors and outdoors)	0.37
	Radon-220 (thoron) (indoors and outdoors)	0.09
	Smoking (Lead-210, Polonium-210, etc.)	0.006*
	Others (uranium, etc.)	0.006
Internal exposure (ingestion)	Mainly Lead-210 and Polonium-210	0.80
	Tritium	0.0000049
	Carbon-14	0.014
	Potassium-40	0.18
Exposure under special environments	Exposure due to hot springs or other subsurface environments	0.005
	Exposure due to the use of aircraft	0.008
<b>Total</b>		<b>2.1</b>

(\* ) Per capita effective doses; The average exposure dose for smokers is 0.040 mSv/y.

# Ground Radiation (World)

Nanograys/h (mSv/y)

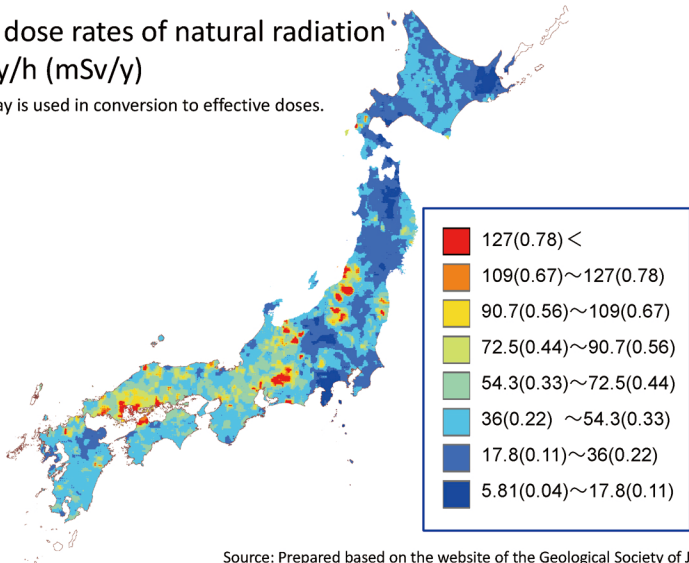
0.7 Sv/year is used in conversion to effective doses.



# Ground Radiation (Japan)

Ambient dose rates of natural radiation  
Nanogray/h (mSv/y)

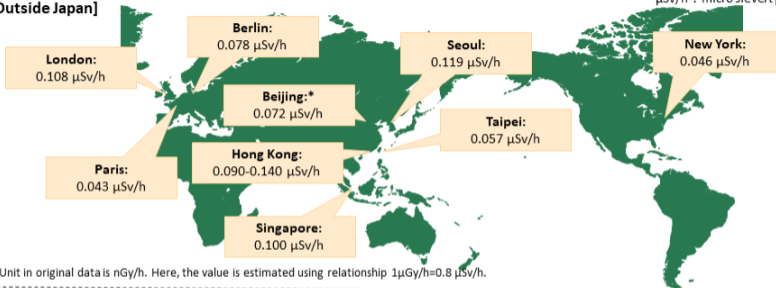
- 0.7 Sv/gray is used in conversion to effective doses.



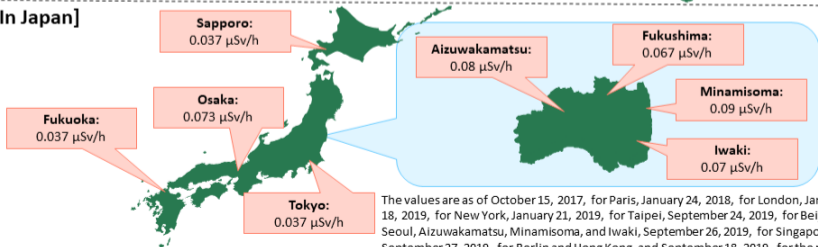
## Results of the Measurements of Ambient Dose Rates in Major Cities

 $\mu\text{Sv/h}$  : micro sievert per hour

## [Outside Japan]

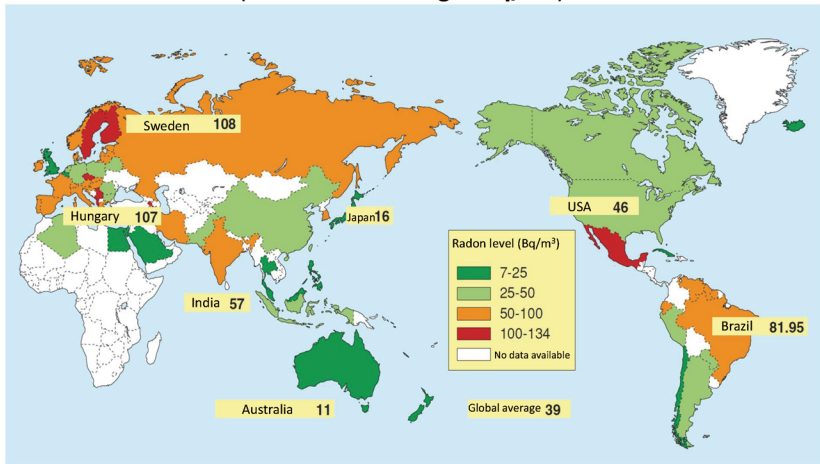


## [In Japan]



# Indoor Radon

Regional differences in exposure from indoor radon  
(arithmetic average: **Bq/m<sup>3</sup>**)



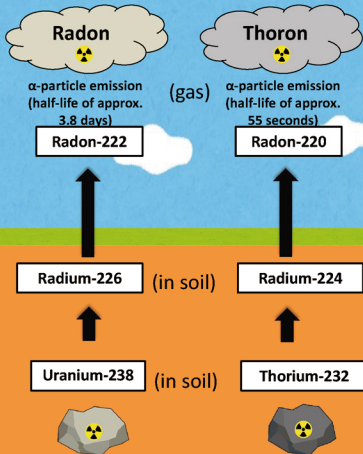
Bq/m<sup>3</sup>: becquerels/cubic meter

Source: Prepared based on the 2006 UNSCEAR (United Nations Scientific Committee on the Effects of Atomic Radiation) Report

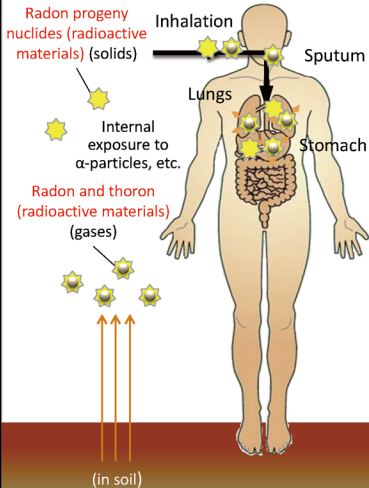


# Internal Exposure to Radon and Thoron through Inhalation

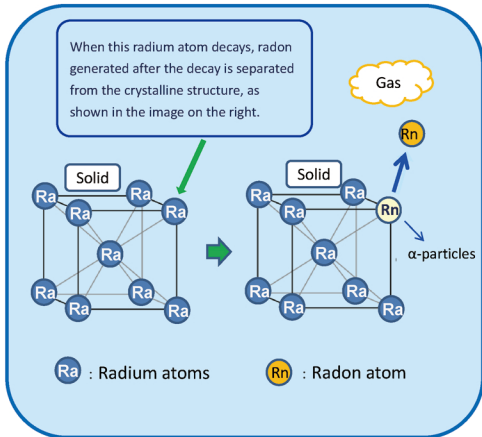
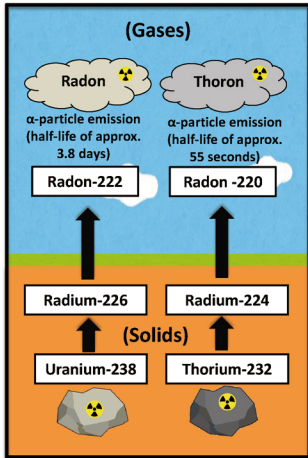
Generation of radon and thoron  
(transfer into the air)



Internal exposure to radon, thoron, and progeny nuclides



It may seem strange that solid radium directly turns into radon gas. This is caused by radioactive decay that causes atoms to change.



# Natural Radioactive Materials in the Body and Foods

## Radioactive materials in the body



When body weight is 60kg

Potassium-40	※ 1	4,000Bq
Carbon-14	※ 2	2,500Bq
Rubidium-87	※ 1	500Bq
Tritium	※ 2	100Bq
Lead and polonium	※ 3	20Bq

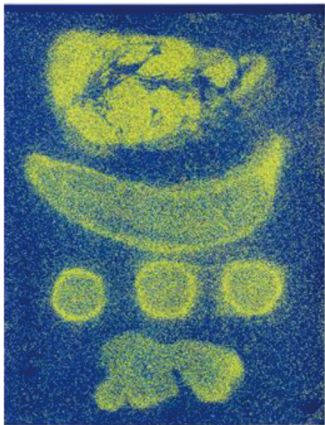
- ※ 1 Nuclides originating from the Earth
- ※ 2 Nuclides derived from N-14 originating from cosmic rays
- ※ 3 Nuclides of the uranium series originating from the Earth

## Radioactivity concentrations (Potassium-40) in foods



Rice: 30; Milk: 50; Beef: 100; Fish: 100; Dry milk: 200; Spinach: 200;  
Potato chips: 400; Green tea: 600; Dried *shiitake*: 700; Dried kelp: 2,000 (Bq/kg)

Bq: becquerels    Bq/kg: becquerels/kilogram



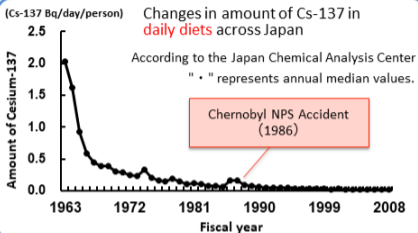
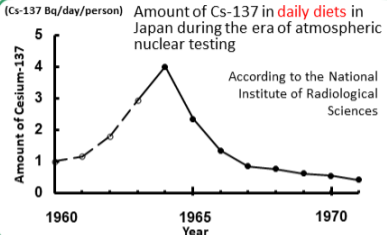
Radiographs of pork meat, banana (cut vertically and horizontally), and ginger

## Radiation from foods

- Mostly  $\beta$ -particles from Potassium-40
- The natural abundance ratio of Potassium-40\* is **0.012%**.
- Potassium-40 has a half-life of  **$1.26 \times 10^9$**  years.

\*Percentage of Potassium-40 relative to the total amount of potassium found in nature

# Changes in Cesium-137 Concentrations in Foods over Time since before the Accident



\*The two studies differ in sampling time and location.



- If an adult keeps consuming the typical diet of the 1960s for a year, internal radiation dose due to Cesium-137 is:

$$4.0 \times 365 \times 0.013 = 19 \mu\text{Sv/y}$$

(Bq/day) (day/year) ( $\mu\text{Sv/Bq}$ )

$$= \underline{0.019 \mu\text{Sv/y}}$$

- (Japanese average)

Annual internal exposure dose due to natural radiation in foods is:

$$\underline{0.99 \text{ mSv/y}}$$

# Radiation Doses from Medical Diagnosis

Type of examination	Diagnostic reference levels <sup>*1</sup>	Actual exposure dose <sup>*2</sup>	
		Dose	Type of dose
General imaging: Front chest	0.4 mGy (less than 100 kV)	0.06 mSv	Effective dose
Mammography (mean glandular dose)	2.4 mGy	Around 2 mGy	Equivalent dose (Mean glandular dose)
Fluoroscopy	IVR (Interventional Radiology): Equipment reference fluoroscopic dose rate 17 mGy/min	Gastric fluoroscopy: 10 mSv/min (25 to 190 sec, varies depending on operators and subjects) <sup>*3</sup>	Effective dose
Dental imaging (Intraoral radiography)	From 1.0 mGy at the frontal teeth of the mandible to 2.0 mGy at the molar teeth of the maxilla (In either case, incident air kerma (Ka,i) [mGy] is measured)	Around 2 - 10 $\mu$ Sv	Effective dose
X-ray CT scan	Adult head simple routine: 77 mGy (CTDIvol)	Around 5 - 30 mSv	Effective dose
	Child (age 5 - 9), head: 55 mGy (CTDIvol)		
Nuclear scanning	Value for each radioactive medicine	Around 0.5 - 15 mSv	Effective dose
PET scan	Value for each radioactive medicine	Around 2 - 20 mSv	Effective dose

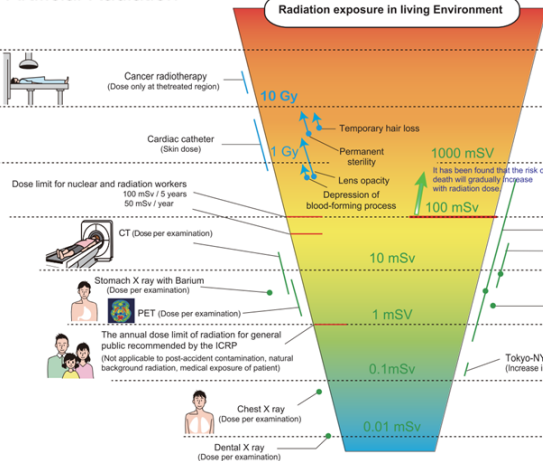
\*1: "National Diagnostic Reference Levels in Japan (2020) (Japan DRLs 2020)," J-RIME, July 3, 2020 (partially updated on August 31, 2020) (<http://www.radher.jp/J-RIME/>)

\*2: "Q&A on Medical Exposure Risks and Protection Regarding Medical Exposure from CT Scans, etc.," National Institutes for Quantum and Radiological Science and Technology (<https://www.qst.go.jp/site/qms/1889.html>)

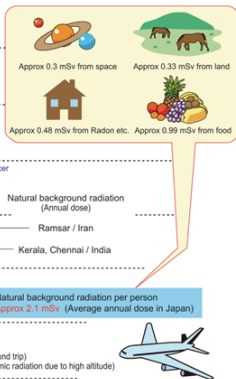
\*3: "Gastric Fluoroscopy" in "X-ray Medical Checkup" in "Basic Knowledge on Medical Radiation," Kitasato University Hospital, Radiology Department  
Prepared based on materials \*1, \*2 and \*3 above

# Comparison of Exposure Doses (Simplified Chart)

## Artificial Radiation



## Natural Background Radiation



### Sources:

- The 2008 UNSCEAR (United Nations Scientific Committee on the Effects of Atomic Radiation) Report
- The 2007 ICRP (International Commission on Radiological Protection) Report
- The exposure guideline of the Japan Association of Radiological Technologists
- "Environmental Radiation in Daily Life (Calculation of the National Doses)," new edition

Prepared by the National Institute of Radiological Sciences, National Institutes for Quantum Science and Technology, based on the sources above (May 2021)

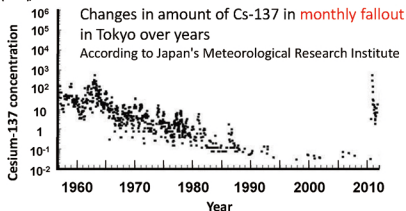
mSv: millisieverts

# Effects of Radioactive Fallout due to Atmospheric Nuclear Testing

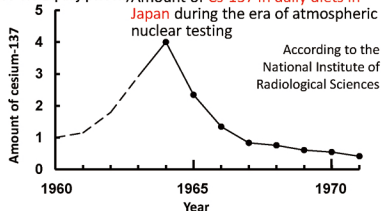
Internal radioactivity: Body weight: 60 kg

Potassium-40: 4,000 Bq; Carbon-14: 2,500 Bq; Rubidium-87: 520 Bq; Tritium: 100 Bq

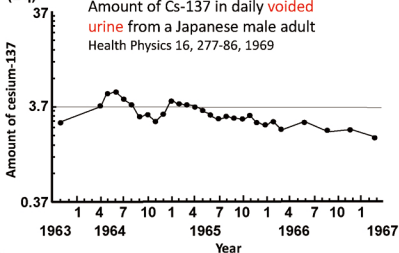
(MBq/km<sup>2</sup> · month)



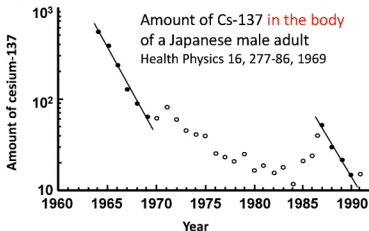
(Cs-137 Bq/day/person) Amount of Cs-137 in daily diets in Japan during the era of atmospheric nuclear testing



(Bq)



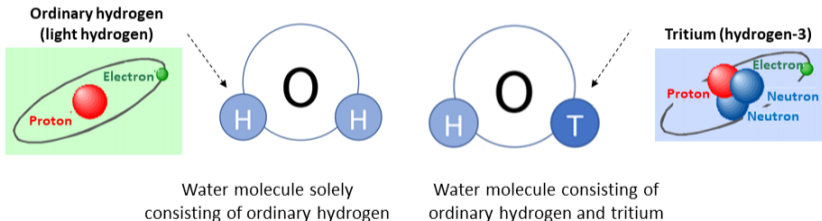
(Bq)





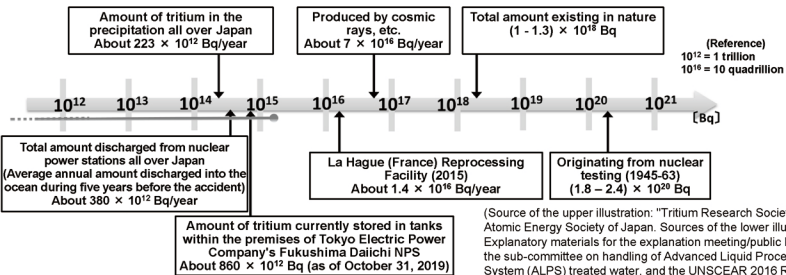
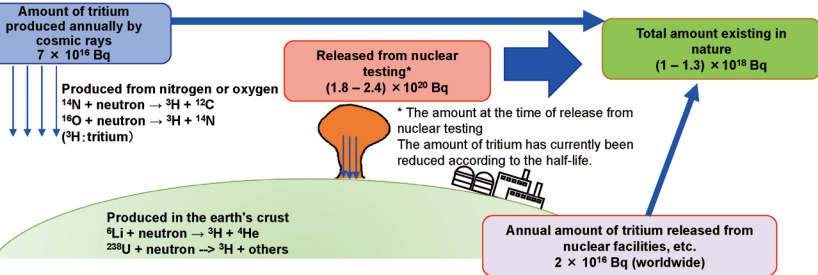
Tritium is a radioisotope of hydrogen, called "hydrogen-3," and exists around us mostly being contained in water molecules.  $\beta$ -particles emitted from tritium only have low energy (18.6 keV at the largest) and can be shielded with a sheet of paper.

## 【Structure of ordinary water molecules】



Source: Prepared based on the "Important Stories on Decommissioning 2018" by the Agency for Natural Resources and Energy, METI, the "Tritiated Water Task Force Report" by the Tritiated Water Task Force (2016), and the "Scientific Characteristics of Tritium (draft)" by the Subcommittee on Handling of the ALPS Treated Water

# Amount of Tritium Existing in Nature



(Source of the upper illustration: "Tritium Research Society" 2014 by Atomic Energy Society of Japan. Sources of the lower illustration: Explanatory materials for the explanation meeting/public hearing by the sub-committee on handling of Advanced Liquid Processing System (ALPS) treated water, and the UNSCEAR 2016 Report)

# Changes in Tritium in Radioactive Fallout over Time

