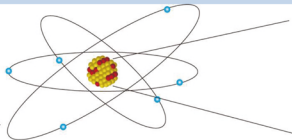


# Where does Radiation Come from?

Material

Atom

Nucleus



Protons

Neutrons



Nucleus



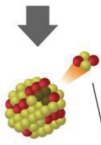
Nucleus



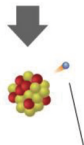
Nucleus  
(high-energy state)



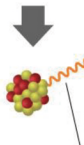
Nucleus



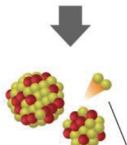
$\alpha$  (Alpha)-particles



$\beta$  (Beta)-particles  
(electrons)



$\gamma$  (Gamma)-rays



Neutron rays

\*X-rays are generated outside a nucleus.

Example of  $\alpha$ -disintegration



Radium-226  
{ 88 protons  
138 neutrons



$\alpha$ -particles  
{ Two protons  
Two neutrons



Radon-222  
{ 86 protons  
136 neutrons

Example of  $\beta$ -disintegration



Tritium  
{ One proton  
Two neutrons



Electrons  
( $\beta$ -particles)



Helium-3  
{ Two protons  
One neutron

# Types of Radiation

## Ionizing radiation

### Particle beams

#### Charged-particle beams (directly ionizing radiation)

- $\alpha$ -particles** (helium nuclei ejected from a nucleus)
- $\beta$ -particles** (electrons ejected from a nucleus), electron beams, positron beams
- Proton beams, deuteron beams, triton beams, heavy-ion beams
- Charged meson beams
- Fission fragments, etc.

#### Uncharged particle beams (indirectly ionizing radiation)

- Uncharged meson beams
- Neutrino
- Neutron beams, etc. (produced in nuclear reactors, accelerators, etc.)

### Electromagnetic waves (indirectly ionizing radiation)

- X-rays** (generated outside a nucleus)
- $\gamma$ -rays** (emitted from a nucleus)

Nonionizing radiation ————— Electric waves, microwaves, infrared rays, visible rays, ultraviolet rays, etc.

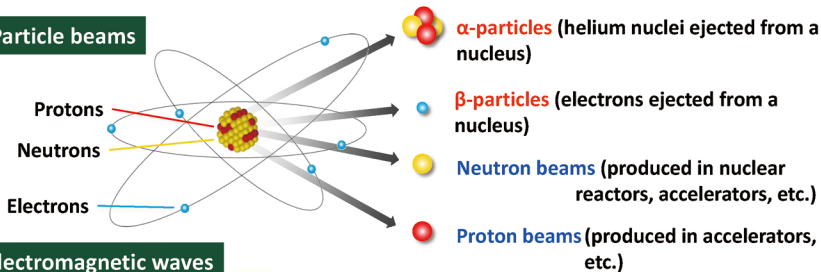
While radiation includes ionizing radiation and nonionizing radiation, radiation usually means ionizing radiation.

# Types of Ionizing Radiation

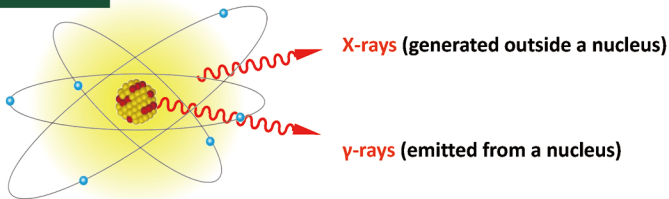
Ionizing radiation

Radiation that causes ionization

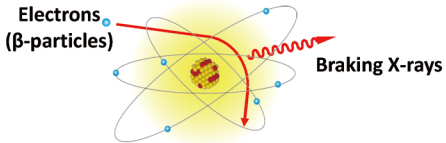
Particle beams



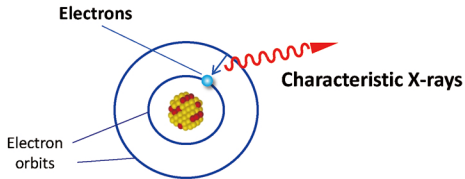
Electromagnetic waves



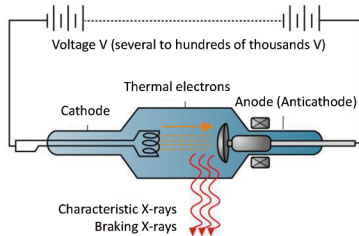
## Braking X-rays



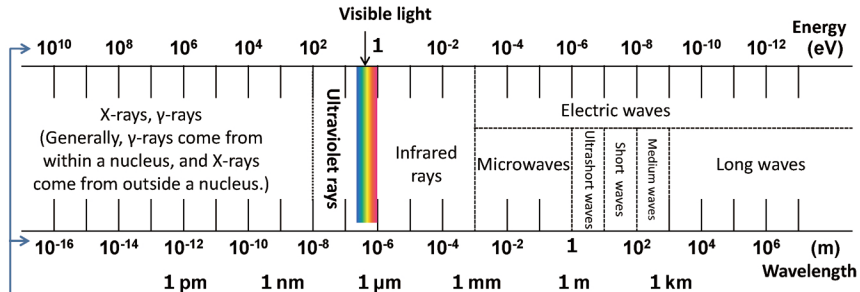
## Characteristic X-rays



## Structural drawing of an X-ray generator



# Types of Electromagnetic Waves

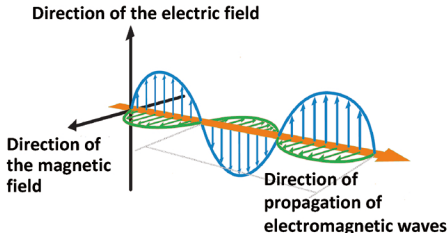


- Light has particle properties in addition to wave properties.
- Electromagnetic waves are called "photons" when they are considered as particles.

The values indicated above show photons' energy (eV) and those indicated below show their wavelengths (m) as wave motions.

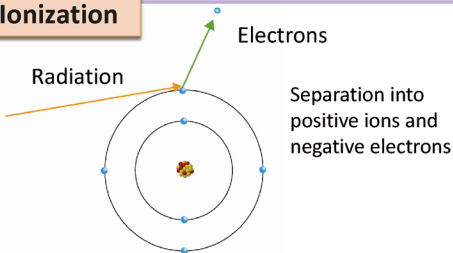
pm: picometers  
nm: nanometers

$\mu$ m: micrometers  
eV: electron volts

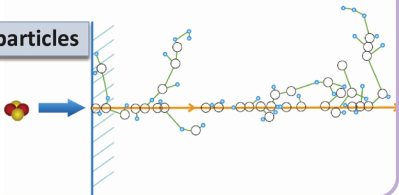


# Ionization of Radiation - Property of Ionizing Radiation

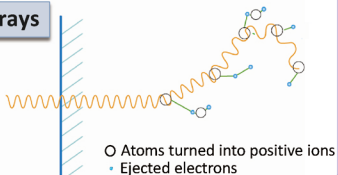
## Ionization



## $\alpha$ -particles



## $\gamma$ -rays



# Types of Radiation and Biological Effects

## • $\alpha$ -particles

- Two protons plus two neutrons
- Helium (He) nuclei
- Charged particles (2+)



High ionization density



## • $\beta$ -particles

- Electrons (or positrons)
- Charged particles (- or +)



Low ionization density



## • $\gamma$ -rays and X-rays

- Electromagnetic waves (photons)



Low ionization density/high penetrating power

## • Neutron beams

- Neutrons
- Uncharged particles



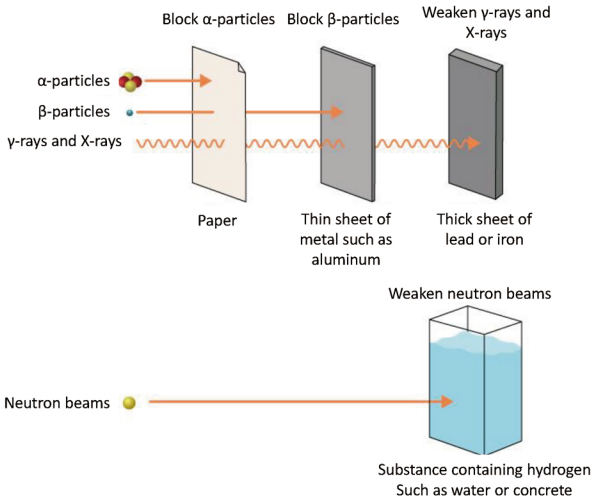
High ionization density



When the ionization number is the same, the higher the ionization density is, the larger the biological effects are.

# Penetrating Power of Radiation

Radiation can be blocked by various substances.





Distance traveling  
in the air

Several centimeters



Several meters  
(depending on the amount  
of energy)



Several tens of  
meters  
(depending on the  
amount of energy)



**$\alpha$ -particles**

Particles (Helium nucleus)  
(One-trillionth of a centimeter)



**$\beta$ -particles**

Particles (electrons)



**$\gamma$ -rays**

**X-rays**



Upon collision with  
the body

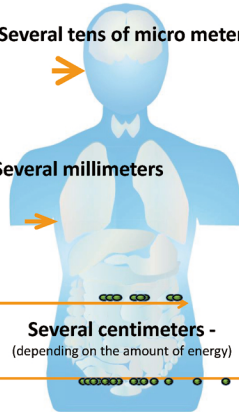
Several tens of micro meters



Several millimeters

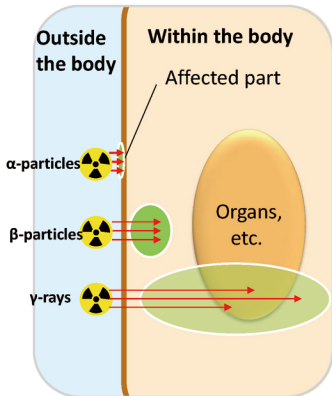


Several centimeters -  
(depending on the amount of energy)



# Penetrating Power and Range of Effects on the Human Body

When radioactive materials are located outside the body



When radioactive materials are located within the body

