

Every year, a large number of reports on research concerning radiation sources and effects are publicized by researchers worldwide.

The United Nations Scientific Committee on the Effects of Atomic Radiation (UNSCEAR) comprehensively evaluates wide-ranging research outcomes, compiles scientific consensus obtained internationally from a politically neutral standpoint, and periodically releases its positions in the form of a report.

The International Commission on Radiological Protection (ICRP) makes recommendations concerning radiological protection frameworks, while referring to reports, etc. by the UNSCEAR. In consideration of ICRP Recommendations and the International Basic Safety Standards established by the International Atomic Energy Agency (IAEA) based on an international consensus, the government of Japan has also formulated laws, regulations and guidelines, etc. concerning radiological protection.

Principles of Radiological Protection

### International Commission on Radiological Protection (ICRP)

#### International Commission on Radiological Protection (ICRP)

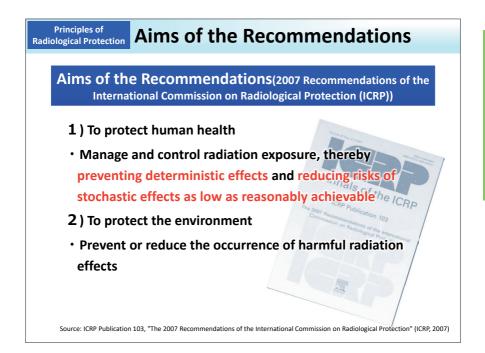
The Commission aims to make recommendations concerning basic frameworks for radiological protection and protection standards. The Commission consists of the Main Commission and five standing Committees (radiation effects, doses from radiation exposures, protection in medicine, application of the Commission's recommendations, and protection of the environment).

(Reference) Dose limits excerpted from ICRP Recommendations

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		1977 Recommendations	1990 Recommendations	2007 Recommendations	2007 Recommendations
	Dose limits (occupational exposure)	50 mSv/year	100 mSv/5 years and 50 mSv/year	100 mSv/5 years and 50 mSv/year	1990 Recommen- dations
	Dose limits (public exposure)	5 mSv/year	1 mSv/year	1 mSv/year	1977 Recommendations
mS۱	r: millisieverts				

The International X-ray and Radium Protection Committee was established in 1928 for the purpose of protecting healthcare workers from radiation hazards. In 1950, the Committee was reorganized into the International Commission on Radiological Protection (ICRP), which was assigned a significant role as an international organization that makes recommendations concerning basic frameworks for radiological protection and protection standards. In recent years, the Commission made recommendations in 1977, 1990 and 2007 (p.155 of Vol. 1, "Aims of the Recommendations"). When the ICRP releases its recommendations, many countries review their laws and regulations on radiological protection accordingly (p.165 of Vol. 1, "ICRP Recommendations and Responses of the Japanese Government").

ICRP Recommendations are based on wide-ranging scientific knowledge, such as that obtained through epidemiological studies on atomic bomb survivors, and its radiological protection system has been maintained since 1990 on the basis of its position that comprehensive estimation of deterministic effects and stochastic risks is basically unchanged.



The ICRP makes recommendations with the aim of contributing to an appropriate level of protection of human beings and the environment against the detrimental effects of ionizing radiation exposure without unduly limiting preferable human beings' efforts and behavior associated with the use of radiation.

The 2007 Recommendations state that in order to achieve this, scientific knowledge on radiation exposure and its health effects is an indispensable prerequisite, but due consideration needs to be given to social and economic aspects of radiological protection in the same manner as in other risk management-related sectors.

The major aim of the ICRP Recommendations has been the protection of human health, but the aim to protect the environment was newly added in the 2007 Recommendations.

Principles of Radiological Protection

People's exposure to radiation

#### **Exposure Situations and Protection Measures**

Planned exposure situations	Existing exposure situations	Emergency exposure situations
Situations where protection measures can be planned in advance and the level and range of exposure can be reasonably forecast <b>Dose limits</b> (Public exposure) 1 mSv/year (Occupational exposure) 100 mSv/5 years and 50mSv/year <b>Measures</b> Manage disposal of radioactive waste and long-lived radioactive waste	Situations where exposure has already occurred as of the time when a decision on control is made Reference level A lower dose range within 1 to 20 mSv/year, with a long-term goal of 1 mSv/year Measures Ensure voluntary efforts for radiological protection and cultivate a culture for radiological protection	Contingency situations where urgent and long-term protection measures may be required <b>Reference level</b> Within 20 to 100 mSv/year <b>Measures</b> Evacuate, shelter indoors, analyze and ascertain radiological situations, prepare monitoring, conduct health examinations, manage foods, etc.
So mSv: millisieverts	urce: ICRP Publication 103, "The 2007 Recomme Radiological Protection" (ICRP, 2007)	ndations of the International Commission on

The International Commission on Radiological Protection (ICRP) categorizes exposure situations into normal times that allow planned control (planned exposure situations), emergencies such as an accident or nuclear terrorism (emergency exposure situations), and the recovery and reconstruction period after an accident (existing exposure situations) and sets up protection standards for each of them.

In normal times, protection measures should aim to prevent any exposure that may cause physical disorders and to reduce risks of developing cancer in the future as low as possible. Therefore, the dose limit for public exposure is set at 1 mSv per year, requiring proper management of places where radiation or radioactive materials are handled to ensure that annual public exposure doses do not exceed this level.

For workers who handle radiation, the dose limit is set at 100 mSv per five years.

On the other hand, in an emergency such as a nuclear accident (emergency exposure situations), as physical disorders that would never be seen in normal times may develop, priority should be placed on measures to prevent serious physical disorders rather than on measures to be taken in normal times (to reduce risks of developing cancer in the future). Therefore, a reference level of 20 to 100 mSv/year is set for the public instead of applying dose limits and efforts to reduce exposure doses are required. For people who are engaged in emergency measures or rescue activities, a level of 500 to 1,000 mSv may sometimes be adopted as a rough indication depending on the circumstances.

Then, in the recovery and reconstruction period (existing exposure situations), a reference level is to be set within the range of 1 to 20 mSv/year, which is lower than the reference level in an emergency but higher than the dose limits applicable in normal times. (Related to p.165 of Vol. 1, "ICRP Recommendations and Responses of the Japanese Government")

# Principles of Radiological Protection Biological Aspect Health effects of radiation have deterministic effects and stochastic effects. • Absorbed doses up to approx. 100 mGy are not judged to cause any clinically significant dysfunction in any tissues. • In the range below approx. 100 mSv, the occurrence of stochastic effects is assumed to increase in proportion to increases in equivalent doses in organs and tissues. • (Adoption of the linear non-threshold (LNT) model)

- The dose and dose-rate effectiveness factor for solid cancer is 2.
- Assuming a linear reaction at low doses, the fatality risks due to cancer and hereditary effects increase by approx. 5% per sievert.

Source: ICRP Publication 103, "The 2007 Recommendations of the International Commission on Radiological Protection" (ICRP, 2007)

One of the aims of the ICRP Recommendations is to provide considerations and assumptions for building a radiological protection system, thereby preventing the occurrence of deterministic effects. The ICRP recommends the introduction of protection measures in cases where annual doses have increased close to 100 mGy ( $\doteq$  100 mSv), which is the minimum threshold.

The probability of stochastic effects is very low in the case of annual doses below approx. 100 mSv, and the linear non-threshold (LNT) model, which is based on the assumption that the occurrence of stochastic effects increases in proportion to increases in radiation doses exceeding background doses, is considered to be practical for the management of radiological protection at low doses and low dose rates, and also preferable from the viewpoint of the precautionary principle.

While the ICRP uses, as the grounds for its recommendations, the data for atomic bomb survivors, which is the data concerning a single exposure, what should be controlled is mostly a long-term gradual exposure. Therefore, the ICRP makes adjustments to offset mitigated effects due to low doses and low dose rates. Various values have been reported as a result of animal testing and experiments using human cells to induce chromosomal abnormalities or mutations, but the dose and dose-rate effectiveness factor for radiological protection has been defined as 2. In other words, if the total exposure dose is the same, long-term low-dose exposure would cause half the effects as those caused by exposure at one time.

As a result of the abovementioned adjustments, risks of fatal cancer are considered to increase by approx. 5% per sievert at low doses and low dose rates.

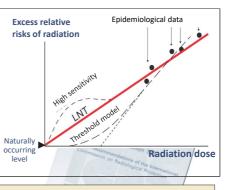
(Related to p.80 of Vol. 1, "Deterministic Effects and Stochastic Effects")

Principles of Radiological Protection

## **Disputes over the LNT Model**

- Affirmative positions: National Academy of Sciences (2006) There is no specific safety dose for radiation exposure.
- O Critical positions:
  - Académie de Médecine; Académie de Science (2005)

Exposure to radiation below a certain dose does not actually cause cancer, leukemia, etc. and therefore, the LNT model represents overestimation not suited to the reality.



### ⇒ The International Commission on Radiological Protection (ICRP) adopts the linear non-threshold (LNT) model as a simple and reasonable assumption for the purpose of radiological protection.

Disputes over the appropriateness of adopting the linear non-threshold (LNT) model for the evaluation of risks of stochastic effects for radiation below 100 mSv have not been settled scientifically. For example, in 2006, the National Academy of Sciences (NAS) publicized its position that the LNT model is scientifically appropriate, stating that there is epidemiological evidence to prove that radiation below 100 mSv also increases cancer risks.

On the other hand, the Académie de Médecine and the Académie de Science jointly publicized their position in 2005, stating that exposure to radiation below a certain dose does not actually cause cancer, leukemia, etc. and therefore that the LNT model represents overestimation not suited to the reality. As the grounds for their position, they cited such facts as that increases in cancer risks are not observed in data for residents in high natural radiation areas in India and China and that defensive biological reactions against low-dose radiation have been found one after another.

The ICRP Recommendations are intended to achieve a practical aim of radiological protection, i.e., the provision of a simpler and more reasonable assumption for the management of risks of lowdose exposure, by adopting the LNT model and defining the dose and dose-rate effectiveness factor as 2. On the other hand, the Recommendations also state that it is judged inappropriate for public health planning to estimate hypothetical incidences of cancer or hereditary diseases among a large number of people due to long-term exposure to very low doses of radiation in consideration of the uncertainties concerning low-dose exposure.

(Related to p.80 of Vol. 1, "Deterministic Effects and Stochastic Effects")

\*Source:

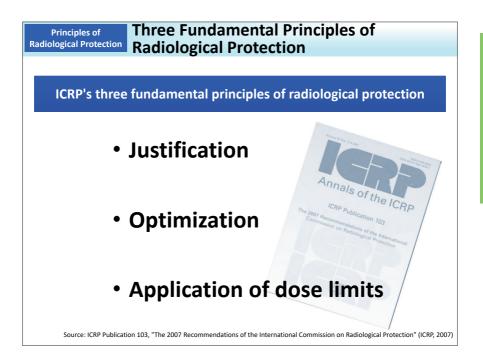
The National Academy of Sciences, "Health Risks from Exposure to Low Levels of Ionizing Radiation: BEIR VII Phase 2," 2006

 Aurengo, A. et al., "Dose-effect relationships and estimation of the carcinogenic effects of low doses of ionizing radiation," Académie des Sciences - Académie nationale de Médecine, 2005
 ICRP Publication 103, "The 2007 Recommendations of the International Commission on

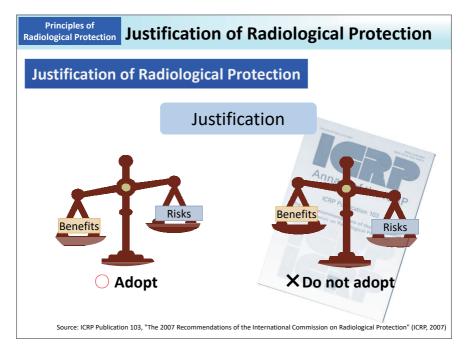
 ICRP Publication 103, "The 2007 Recommendations of the International Commission on Radiological Protection" (ICRP, 2007)

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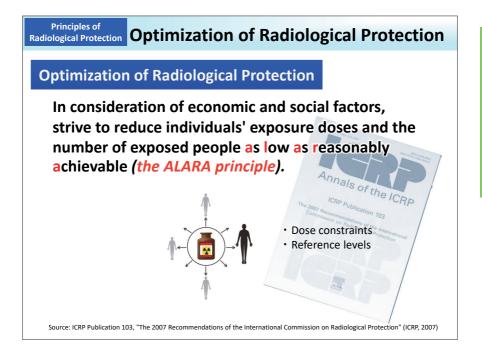


In cases of cancer and hereditary effects, effects appear stochastically. At present, the linear non-threshold (LNT) model is adopted in radiological protection even for low doses (p.158 of Vol. 1, "Disputes over the LNT Model"), due to which the safety and the danger cannot be clearly divided. Therefore, the protection level is considered based on the idea that risks cannot be completely eliminated and on an assumption that such risks can be tolerated. This is the very basis of the principles of radiological protection, placing emphasis on the "justification," "optimization" and "application of dose limits" (p.160 of Vol. 1, "Justification of Radiological Protection," p.161 of Vol. 1, "Optimization of Radiological Protection," and p.163 of Vol. 1 "Application of Dose Limits").



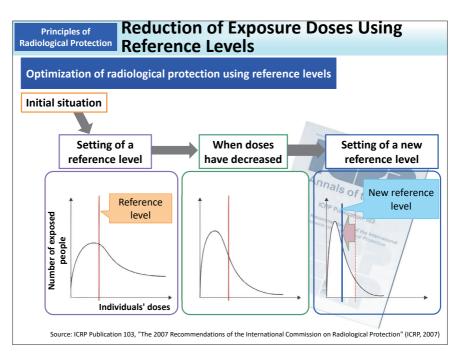
The first principle is the justification of radiological protection. This is the fundamental principle that an act of using radiation is permitted only when the benefits or merits outweigh the radiation risks.

This principle is applied not only to acts of using radiation but also to all activities that bring about changes in exposure situations. In other words, this is also applied to emergency exposure situations and existing exposure situations, as well as to planned exposure situations. For example, justification is required even in the case of considering decontamination of contaminated areas.



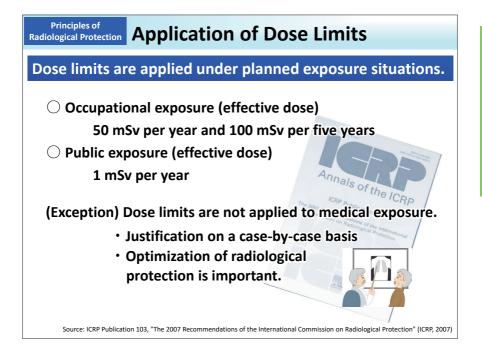
The second principle is the optimization of radiological protection. When merits of an act of using radiation outweigh radiation risks, it is decided to use radiation by taking measures to reduce exposure doses as low as reasonably achievable. This is called the ALARA principle. The optimization of radiological protection means to strive to reduce exposure doses as low as possible, while taking into consideration social and economic balances, and does not necessarily mean to minimize exposure doses.

In order to promote the optimization of radiological protection, dose constraints and reference levels are utilized. Reference levels are adopted as indicators to limit individuals' doses from specific radiation sources in decontamination work, for example.



The concept of reference levels as suggested in the 2007 Recommendations of the ICRP has been adopted in promoting measures to reasonably reduce exposure doses due to the accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS. In an emergency such as an accident or nuclear terrorism (emergency exposure situations), the focus is placed on measures to prevent serious physical disorders. Therefore, dose limits (limits for exposure to all regulated radiation sources under planned exposure situations) are not applied. Instead, a reference level is set within the range of annual doses of 20 to 100 mSv for the public and protection activities are carried out so as to limit individuals' doses below that level. Physical disorders that would never be seen in normal times may develop in an emergency. Accordingly, measures to prevent such physical disorders are prioritized over measures to be taken in normal times (to reduce risks of developing cancer in the future).

Reference levels aim to ensure that no one receives an unduly high dose in a circumstance where exposure doses among individuals are not even. When considering protection measures for the entirety, if there are people who are likely to receive doses exceeding the predetermined reference level, countermeasures for those people are preferentially taken. If dose disparity within a group diminishes as a result of such intensive countermeasures, and there is almost no one who receives a high dose exceeding the reference level, a new lower reference level is set as necessary to further reduce exposure doses as a whole. In this manner, exposure dose reduction can be achieved efficiently by setting appropriate reference levels depending on the circumstances.



The third principle of radiological protection is the application of dose limits. The 2007 Recommendations of the ICRP specify the effective dose limit for occupational exposure (excluding radiation work in an emergency) as 100 mSv per five years and 50 mSv for the specific one year.

The effective dose limit for public exposure is specified as 1 mSv per year.

Dose limits are the standard limits below which the total exposure to all radiation sources under management is to be controlled. The goal is not to merely keep the total exposure below those dose limits but continued efforts are required to reduce exposure doses through further optimizing radiological protection. In this sense, dose limits do not stand for permissible exposure doses, nor do they represent the threshold to divide the safety and the danger.

Regarding medical exposure in treatment or health checkups, dose limits are not applied. This is because the application of dose limits to medical exposure may hinder patients from receiving necessary inspections or treatment and is sometimes detrimental to them. Accordingly, the justification is to be made from three viewpoints (the fact that radiation use in medicine is more beneficial than harmful to patients; application of specific methods to patients exhibiting specific symptoms; and application of methods customized for respective patients), and doses are to be optimized by applying diagnostic reference levels, etc.

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# **Dose Limits** Comparison between ICRP Recommendations and Domestic Laws and Regulations

	DU	mestic Laws	and Regulation	0115	
		Occupational exposure		Public exposure	
		2007 Recommendations of the ICRP	Laws and regulations concerning the prevention of radiation hazards (Japan), as of March 2012	2007 Recommendations of the ICRP	Laws and regulations concerning the prevention of radiation hazards (Japan), as of March 2012
Effecti	ive dose limits	The average annual dose for the prescribed five years should not exceed 20 mSv and the annual dose for any one year should not exceed 50 mSv.	Same as the Recommendations	1 mSv/year (Exceptionally, if the average annual dose for five years does not exceed 1 mSv, exposure exceeding the limit for a single year may be sometimes permitted.)	Dose limits are not specified, but doses at the boundaries of business establishments, including those due to exhaust gas and water, are regulated not to exceed the dose limit of 1 mSv/year.
its i	Eye lenses	150m Sv/year	150 mSv/year	15 mSv/year	-
Equivalent dose limits	Skin	500 mSv/year	500 mSv/year	50 mSv/year	-
do Eq	Fingers and toes	500 mSv/year	-	-	-
Dose limi female ra workers	e radiation	The equivalent dose of a fetus during gestation after reporting pregnancy (exposure in utero) should not exceed 1 mSv.	5 mSv/3 months Equivalent dose limit for the abdominal surface after coming to know of pregnancy until delivery: 2 mSv Internal exposure: 1 mSv	_	-
mSv: r	millisieverts		Source: Prepared based on the regulations concerning	2007 Recommendations of the the prevention of radiation has	

Present laws and regulations in Japan have not yet incorporated the 2007 Recommendations of the ICRP, but dose limits specified in the 2007 Recommendations are mostly the same as those in the 1990 Recommendations. Therefore, dose limits in Japan also mostly coincide with those specified in the 2007 Recommendations. Japan has uniquely specified dose limits for female radiation workers (5 mSv per three months).

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Dose Limits ICRP Recommendations and Responses of the Japanese Government				
	2007 Recommen	dations of the ICRP	Responses at the time of the accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS	
	Rescue activities (Volunteers who have obtained the relevant information)	When benefits for other people outweigh the rescuers' risks, dose limits are not applied.	Special Provisions of the Ordinance on Prevention of Ionizing Radiation Hazards (Ministry of Health, Labour and Welfare) The emergency exposure dose limit was temporarily raised to <b>250 mSv</b> from the	
Occupationa exposure	Other emergency activities	Up to 500 mSv	conventional level of <b>100 mSv</b> (from March 14 to December 16, 2011). The Ordinance on Prevention of Ionizing Radiation Hazards was partially amended to raise the exceptional emergency dose limit to <b>250 mSv</b> (enforced on April 1, 2016).	
Public	Emergency exposure situations	The limit is to be set within the range of <b>20 to 100</b> <b>mSv</b> /year.	e.g. Standards for evacuation in Deliberate Evacuation Areas: <b>20 mSv</b> /year	
exposure	Reconstruction period (Existing exposure situations)	The limit is to be set within the range of <b>1 to 20</b> <b>mSv</b> /year.	e.g. Additional exposure dose to be achieved in the long term: <b>1 mSv</b> /year	
mSv: millisi	Source: Prepared based on the 2007 Recommendations of the ICRP and the Special Provisions of the Ordina mSv: millisieverts on Prevention of Ionizing Radiation Hazards (Ministry of Health, Labour and Welfare: MHLW)			

The accident at TEPCO's Fukushima Daiichi NPS occurred while deliberations were continuing over the incorporation of the 2007 Recommendations of the ICRP into domestic laws and regulations.

The accident changed exposure situations, and the idea of reference levels, which had been unfamiliar to Japanese laws and regulations, was adopted for public exposure. In exposure dose control using reference levels, an initial reference level is first set based on the standards for respective exposure situations specified in the 2007 Recommendations of the ICRP so as to ensure that no one receives an unduly high dose. Secondly, if the situation has improved and there is almost no one who receives a high dose exceeding the reference level, a new lower reference level is set as necessary to efficiently achieve exposure dose reduction.

In the meantime, regarding occupational exposure, the emergency dose limit was temporarily raised from 100 mSv to 250 mSv as an exception for an unavoidable case for the purpose of preventing the expansion of the disaster at the NPS. Later, as the work to achieve stable cold shut-down conditions of the reactors was completed, this exceptional measure was abandoned.

Considering the need to develop regulations on the prevention of radiation hazards during emergency work in preparation for any possible nuclear emergencies at nuclear facilities in the future, the Ordinance on Prevention of Ionizing Radiation Hazards was partially amended to raise the exceptional emergency dose limit to 250 mSv. The amended Ordinance was put into force on April 1, 2016.

### Dose Limits Comparison of Regulation Values for Foods

Radionuclide	Japan	Codex Alimentarius Commission	EU	US
Radioactive cesium	Milk 50 Infant foods 50 General foods 100	Infant foods 1,000	Milk 1000 Infant foods 400 General foods 1,250	All foods 1,200
Upper limits for additional doses	1mSv	1mSv	1mSv	5mSv
Assumed percentages of foods containing radioactive materials	50%	10%	10%	30%

Indices concerning radioactive materials in foods (Bq/kg)

\* The Codex Alimentarius Commission is an intergovernmental body created in 1963 by the Food and Agriculture Organization of the United Nations (FAQ) and the World Health Organization (WHQ) for the purpose of protecting consumers' health and ensuring fair-trade practices in the food trade, etc.; The Commission establishes international standards for foods.

\* Standard limits incorporate effects of the amount of food intake and assumed percentages of foods containing radioactive materials. Therefore, the values are not suitable for inter-comparison.

\* Indicated standard limits for drinking water are the WHO guidance levels of radioactive materials, which are referred to in respective countries, and standard limits for radioactive materials vary by country due to differences in adopted preconditions. Therefore, the values are not suitable for inter-comparison.

Source: Prepared based on "Food and Radiation Q&A," Consumer Affairs Agency

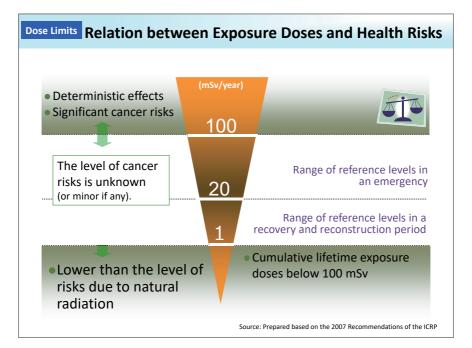
In Japan, new standard limits for radioactive materials in foods were established and were put into force on April 1, 2012. The new standard limits were set by classifying foods into four categories and the standard limit for drinking water, which is most frequently taken by people, was set at 10 Bq/kg.

The standard limit for milk, which children generally drink a lot of, was reduced to 50 Bq/kg. Additionally, a new category, "infant foods," was made for ensuring safety for infants and the standard limit therefor was set at 50 Bq/kg, the same as that for milk. The standard limit for other general foods is 100 Bq/kg.

All foods other than infant foods were categorized as general foods based on the idea to minimize gaps in additional doses caused by differences in individuals' eating habits. The value was set with sufficient room to ensure safety no matter what foods people eat as long as radioactive Cs concentrations therein are within the standard limit.

Regulation values vary by country due to differences in annual exposure dose limits based on which the respective countries set their standard limits and in contamination rates in foods, etc. (In Japan, regulation values were set on the safe side based on the annual exposure dose limit of 1 mSv and on the assumption that 50% of general foods and 100% of milk and infant foods are contaminated. On the other hand, the Codex Alimentarius Commission specifies the annual exposure dose limit as 1 mSv and assumes that 10% of foods are contaminated.)

(Related to p.43 of Vol. 2, "Standard Limits Applied from April 2012")

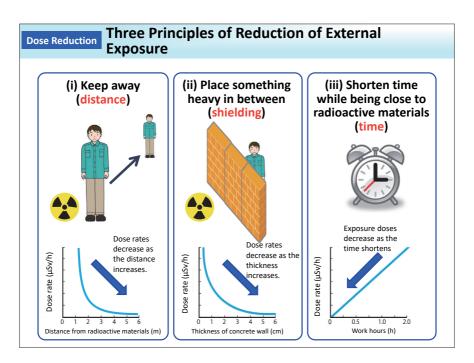


4.2 Dose Limits

There is scientific evidence for the fact that radiation doses of 100 to 200 mSv or over increase cancer risks. Therefore, in an emergency due to a radiation accident, the initial reference level is set to avoid annual exposure doses of 100 mSv or over in order to prevent serious physical disorders. When the situation improves as the accident is brought under control and there is almost no one who receives a high dose exceeding the initial reference level, a new lower reference level (such as 1 to 20 mSv per year) is set to curb increases in risks of any possible cancer in the future, thereby further promoting exposure dose reduction (p.156 of Vol. 1, "Exposure Situations and Protection Measures").

As the standard limit in normal times, 1 mSv/year is adopted. As a result, some misunderstand that radiation exposure exceeding 1 mSv per year is dangerous or that they may be exposed to radiation up to that level. However, dose limits do not represent the threshold dividing the safety and the danger.

It is not that radiation exposure up to 1 mSv per year is permissible. Principally, radiation exposure should be reduced as low as practically achievable in light of various circumstances.



There are three ways to reduce external exposure doses.

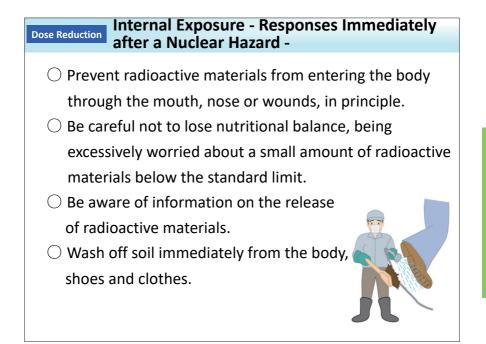
The first is to keep away from radioactive materials such as by removing soil contaminated with radioactive materials and isolate it from people's living environment.

The second is to shield radiation such as by staying indoors, replacing topsoil contaminated with radioactive materials with subsoil, and using uncontaminated soil as a shielding material.

The third is to shorten the time to stay at places with high ambient dose rates. (Related to p.47 of Vol. 1, "Characteristics of External Exposure Doses")

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4.3 Dose Reduction



As causes of internal exposure, both inhalation and ingestion of foods need to be taken into consideration. For example, when calculating exposure doses for children engaging in outdoor activities at places with high ambient doses, doses due to internal exposure account for only around 2% to 3% and exposure doses are mostly due to radiation from outside of the body. Therefore, people do not have to be too nervous about exposure through inhalation and proper daily hygienic control (taking a bath, getting a haircut, washing hands, cleaning, and doing the laundry, etc.) is effective in reducing internal exposure to some extent.

In the meantime, regarding the possibility of internal exposure caused by ingestion of foods, attention needs to be paid to wild foods or other foods whose safety cannot be confirmed. In particular, special attention is required for ferns and mushrooms, which have a property to concentrate cesium.

Unlike ambient dose rates, internal exposure doses cannot be easily checked personally, so it is recommendable to refer to measurement results, etc. released by relevant ministries and agencies. Data on radioactivity concentrations in foods is released by the Ministry of Health, Labour and Welfare and the Ministry of Agriculture, Forestry and Fisheries.

# Dose Reduction Removal of Radioactive Cesium through Cooking and Processing of Foods

### Radioactive materials can be reduced through cooking.

Leaf vegetables (spinach, etc.)       Washing - Boiling       7~78         Bamboo shoots       Boiling       26~36         Marron       Boiling - Peeling astringent skin       11~34         Japanese radish       Peeling       24~46         Japanese plum       Salting       34~43         Cherry leaves       Salting       78~87         Dried shiitake mushrooms       Reconstituting (do not use soaking water)       51~54         Beef meat       Boiling sliced meat (Shabu-shabu)       55~69			0
Leaf vegetables (spinach, etc.)Washing - Boiling $7 \sim 78$ Bamboo shootsBoiling $26 \sim 36$ MarronBoiling - Peeling astringent skin $11 \sim 34$ Japanese radishPeeling $24 \sim 46$ Japanese plumSalting $34 \sim 43$ Cherry leavesSalting $78 \sim 87$ Dried shiitake mushroomsReconstituting (do not use soaking water) $51 \sim 54$ Beef meatBoiling sliced meat (Shabu-shabu) $55 \sim 69$ FishCooked lake smelt soaked in Japanese sweet and peppery vegetable sauce $22 \sim 32$ Avoid eating wild foods too much. Removal rate (%) = $\left(1 - \frac{Total amount of radioactivity in cooked or processed foods (Bq)}{Total amount of radioactivity in raw materials (Bq)} \right) \times$	ltem	Cooking/Processing methods	Removal rate (%)
(spinach, etc.)       Washing - Boiling $1^{1 \sim 78}$ Bamboo shoots       Boiling $26 \sim 36$ Marron       Boiling - Peeling astringent skin $11 \sim 34$ Japanese radish       Peeling $24 \sim 46$ Japanese plum       Salting $34 \sim 43$ Cherry leaves       Salting $78 \sim 87$ Dried shiitake mushrooms       Reconstituting (do not use soaking water) $51 \sim 54$ Beef meat       Boiling sliced meat (Shabu-shabu) $55 \sim 69$ Fish       Cooked lake smelt soaked in Japanese sweet and peppery vegetable sauce $22 \sim 32$ • Avoid eating wild foods too much. Removal rate (%) = $\left(1 - \frac{\text{Total amount of radioactivity in cooked or processed foods (Bq)}{\text{Total amount of radioactivity in raw materials (Bq)}} \right) \times$	Rice	ice Polishing - Washing - Cooking	
Marron       Boiling - Peeling astringent skin       11~34         Japanese radish       Peeling       24~46         Japanese plum       Salting       34~43         Cherry leaves       Salting       78~87         Dried shiitake mushrooms       Reconstituting (do not use soaking water)       51~54         Beef meat       Boiling sliced meat (Shabu-shabu)       55~69         Fish       Cooked lake smelt soaked in Japanese sweet and peppery vegetable sauce       22~32         • Avoid eating wild foods too much. Removal rate (%) = $\left(1 - \frac{\text{Total amount of radioactivity in cooked or processed foods (Bq)}{\text{Total amount of radioactivity in raw materials (Bq)}}\right) ×   $	Leaf vegetables (spinach, etc.)	Washing - Boiling	
Japanese radish       Peeling       24~46         Japanese plum       Salting       34~43         Cherry leaves       Salting       78~87         Dried shitake mushrooms       Reconstituting (do not use soaking water)       51~54         Beef meat       Boiling sliced meat (Shabu-shabu)       55~69         Fish       Cooked lake smelt soaked in Japanese sweet and peppery vegetable sauce       22~32         • Avoid eating wild foods too much. Removal rate (%) = $\left(1 - \frac{\text{Total amount of radioactivity in cooked or processed foods (Bq)}{\text{Total amount of radioactivity in raw materials (Bq)}}\right) ×   $	Bamboo shoots	Boiling	26~36
Japanese plum       Salting       34~43         Cherry leaves       Salting       78~87         Dried shiitake mushrooms       Reconstituting (do not use soaking water)       51~54         Beef meat       Boiling sliced meat (Shabu-shabu)       55~69         Fish       Cooked lake smelt soaked in Japanese sweet and peppery vegetable sauce       22~32         • Avoid eating wild foods too much. Removal rate (%) = $\left(1 - \frac{\text{Total amount of radioactivity in cooked or processed foods (Bq)}{\text{Total amount of radioactivity in raw materials (Bq)}}\right) ×   $	Marron	Boiling - Peeling astringent skin	11~34
Cherry leaves       Salting       78~87         Dried shiitake mushrooms       Reconstituting (do not use soaking water)       51~54         Beef meat       Boiling sliced meat (Shabu-shabu)       55~69         Fish       Cooked lake smelt soaked in Japanese sweet and peppery vegetable sauce       22~32 <ul> <li>Avoid eating wild foods too much. Removal rate (%) = <math>\left(1 - \frac{\text{Total amount of radioactivity in cooked or processed foods (Bq)}{\text{Total amount of radioactivity in raw materials (Bq)}}\right) ×   </math></li></ul>	Japanese radish	Peeling	24~46
Avoid eating wild foods too much.       Removal rate (%) = (1 - Total amount of radioactivity in cooked or processed foods (Bq))	Japanese plum	Salting	34~43
mushrooms       Reconstituting (do not use soaking water)       517~54         Beef meat       Boiling sliced meat (Shabu-shabu)       55~69         Fish       Cooked lake smelt soaked in Japanese sweet and peppery vegetable sauce       22~32         Avoid eating wild foods too much.       Removal rate (%) = (1 - Total amount of radioactivity in cooked or processed foods (Bq))       ×	Cherry leaves Salting		78~87
Determinant       Doffing street (streat (str			51~54
FISN       peppery vegetable sauce       227.932         Avoid eating wild foods too much.       Removal rate (%) = $\left(1 - \frac{\text{Total amount of radioactivity in cooked or processed foods (Bq)}{\text{Total amount of radioactivity in raw materials (Bq)}}\right) \times$	Beef meat	Boiling sliced meat (Shabu-shabu)	55~69
Removal rate (%) = $\left(1 - \frac{\text{Total amount of radioactivity in cooked or processed foods (Bq)}{\text{Total amount of radioactivity in raw materials (Bq)}}\right) \times $	Fish	Cooked lake smelt soaked in Japanese sweet and peppery vegetable sauce	22~32
	Rem	oval rate (%) = $\left(1 - \frac{\text{Total amount of radioactivity in cooked of}}{\text{Total amount of radioactivity in radioactivity in radioactivity}}\right)$	

urce: "Environmental Parameters Series Expanded Edition (2013): Radionuclide Removal Rates through Cooking and Processing of Foods - Centered on Data on Radioactive Cs Removal Rates in Japan -" (September 2013), Radioactive Waste Management Funding and Research Center

Immediately after the accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS, radioactive materials detected from vegetables were only attached to the surface thereof, and such radioactive materials could be washed off to some extent.

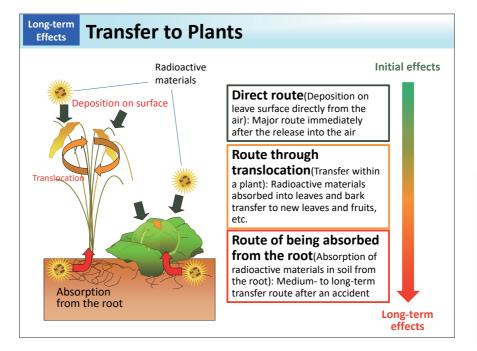
At present, radioactive materials are seldom attached to the surface of vegetables, but some radioactive materials in soil may be taken into vegetables through their roots. However, radioactive cesium adsorbed into vegetables from the roots can be removed through cooking or processing with some ingenuity.

The table above shows removal rates of radioactive cesium in foods.

When boiling vegetables, the longer the boiling time is, the larger the removal rate is. This is considered to be because radioactive cesium in vegetable cells breaks and comes out into the boiling water. Also in the case of salted vegetables, the longer the salting time is, the larger the removal rate is. This is considered to be because radioactive cesium in vegetables is replaced with sodium in salt.

When cooking meat or fish, the amount of radioactive materials can be halved by discarding the cooking liquid. It is known that the removal rate is higher when boiling or cooking than grilling them.

Refer to the webpage (http://www.rwmc.or.jp/library/other/kankyo/, in Japanese) for the details of the related data.

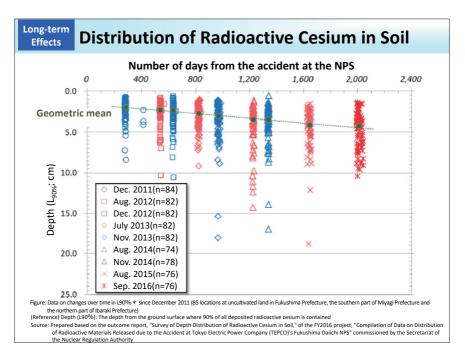


As Cs-137 has a long half-life of 30 years, once released into the environment due to an accident at a nuclear power station or other reasons, its effects may be prolonged. There are roughly three routes through which radioactive materials in the environment transfer to the edible parts of crops.

The first is the route wherein radioactive materials adhere to the surface of edible parts of crops directly from the air. Radioactive materials measured immediately after the accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS were those that were released into the air due to the accident and directly adhered to leaf surfaces.

The second is the route through translocation. Translocation refers to the phenomenon wherein absorbed nutrients or metabolites produced by photosynthesis are transported from some tissue to another tissue in a plant. Radioactive materials that adhere to leaves or bark are sometimes absorbed and transfer to new leaves and fruits within a plant. Relatively high levels of radioactive materials detected in tea leaves, bamboo shoots, loquats, plums, etc. are considered to have followed this route.

The third is the route wherein radioactive materials in soil are absorbed from the root. After the release of radioactive materials into the air stops, radioactive materials that fell onto farmland will mainly follow this route and will be absorbed into crops from the root.



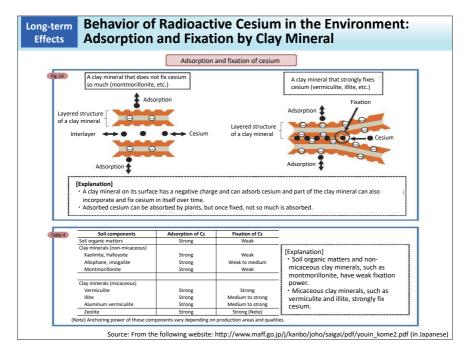
Surveys concerning the depth distribution in soil of radioactive cesium released due to the accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS have been conducted since FY2011 in Fukushima Prefecture, the southern part of Miyagi Prefecture and the northern part of Ibaraki Prefecture.

The depth from the ground surface containing 90% of all deposited radioactive cesium has increased gradually over time, but the geometric mean as of September 2016 was less than 5 cm at 4.2 cm.

Distribution of radioactive cesium varies depending on the status of soil such as cracks and as a result of decontamination work or deep plowing. Clayey soil contains clay minerals such as vermiculite, which strongly adsorb cesium. Cesium adsorbed in such clayey soil becomes hardly soluble in water and is fixed and retained near the surface layer of the soil for a long term.

Accordingly, radioactive cesium thus retained near the surface layer is physically isolated from the root of the types of plants that take root deeper in the soil.

The survey on effects of the Chernobyl nuclear reactor accident that occurred in 1986 revealed that approx. 80% of Cs-137 deposited on soil due to the accident had been staying within 10 cm from the ground surface even after 14 years from the accident (Report of the Chernobyl Forum Expert Group (2006), International Atomic Energy Agency).



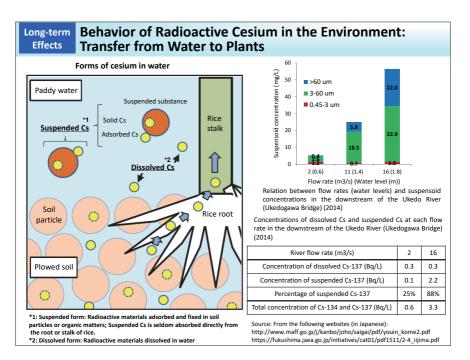
Cesium has a similar chemical property as potassium, etc. (having a positive charge) and can be easily adsorbed by clay minerals that have a negative charge superficially. Furthermore, some clay minerals have the ability to fix cesium that they have adsorbed, as time proceeds. It is known that cesium, once fixed, becomes hardly soluble in water.

Radioactive cesium released into the environment due to the accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS has been adsorbed and fixed by clay minerals in soil as time passes and not much has been absorbed into crops (the above figure).

In particular, micaceous clay minerals, such as vermiculite and illite, are known to have the property to strongly fix cesium (lower table).

Research and studies conducted so far have confirmed a declining trend over time in the concentration of radioactive cesium in river water samples collected in Fukushima Prefecture, as well as a declining trend over time in the concentration of radioactive cesium that flows into rivers from forests, etc.

\* Source: Outcome report of the FY2014 project, "Compilation of Data on Distribution of Radioactive Materials Released due to the Accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS and Development of Transfer Model" commissioned by the Secretariat of the Nuclear Regulation Authority



When paddy fields are plowed and watered, the water contains dissolved cesium and suspended cesium adhering to soil particles, etc. However, cesium adsorbed or fixed in soil is seldom dissolved in water and suspended Cs is not absorbed directly from the root or stalk of rice (figure on the left).

Cesium in reservoirs and water channels is adsorbed or fixed in soil as time passes. Therefore, in surveys in Fukushima Prefecture, radioactive cesium was mostly detected as being dissolved in water under circumstances where the river flow rate and turbidity were low and detected concentrations were lower than the detection limit for ordinary measurements of radioactivity concentrations (approx. 1 Bq/L).

As shown in the upper right figure, when the river flow rate is high such as upon a heavy rain (high water level), the suspensoid concentration also becomes high, but suspended substances strongly adsorb radioactive cesium (suspended Cs). Accordingly, when the water level is high, the concentration of dissolved Cs stays almost the same and only the concentration of suspended Cs becomes higher, but the latter also decreases over time. As the river flow rate increases, particles of suspended substances become larger and the turbidity increases. However, such turbidity can be solved through filtration. As shown in the lower right table, the survey conducted at the Ukedo River in Fukushima Prefecture confirmed that radioactive Cs concentrations in normal times were below the standard limit for drinking water (10 Bq/L) and that radioactive Cs concentrations after filtration were below the detection limit (approx. 1 Bq/L) even for river water with high turbidity collected when the water level is high.

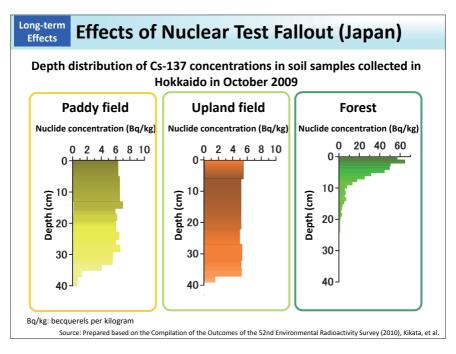
#### Behavior of Radioactive Cesium in the Environment: Long-term **Outflow from Forest Soil** Effects Surveys conducted so far revealed that the annual outflow rate of Cs-137 from forest soil is around 0.02% to 0.3% of the total amount of Cs-137 deposited on nearby watershed soil. [Table 1] Outflow of radioactive Cs from watershed areas to rivers (Outflow rates) Mt. Tsukuba Watershed area Upstream of the Survey period 44 to 45 days\*3 21 months 15 months Amount of Cs-137 deposited on soil 544 298 916 13 170-230 (kBq/m<sup>3</sup>) Amount of outflow of Cs-137\*4 0.087 0.026 0.021 0.06 0.22-0.34 (kBq/m<sup>3</sup>) Percentage of the amount of Cs-137 0.016% 0.009% 0.002% 0.5% 0.12-0.15% outflow against the total amount of Cs-137 deposited on soil Percentage of the annual amount of 0.13% 0.07% 0.02% 0.26% 0.10-0.12% outflow of Cs-137\*5 11: Source) Outcome report of the PD312 commissioned radiation measurement project, "Establishment of Methods to Accertain Long term Effects of Radioactive Materials Rei due to the Accelerate at Totopic Entric Poer Company (FECO) Sistabilities Datich NG5<sup>+</sup>, MAA 22: (Source) National Institute for Invironmental Studies, 2012 and 2013 31: Estrated and totaled companible data for the tert we avertained areas obtained from October 1 to 9 or 10, from October 22 to November 3, and from November 29 or 30 to December 18 or 19, 2012 (44 to 45 days) \*4: O Watershed areas around Mt. Iboishi, Mt. Ishihira and Mt. Kodaishi: Total amount of Cs-137 in river water (dissolved Cs-137, suspended substances (SS) and large orga C) wateries areas adout and incomin, which domina mit. Notabani: (bital amount of C+12 in Intervieter (BoxOverG C+13 r, Subpercies Substances (bit) and urge organic matter (Bevers and branches) (Bowing) in the river). Biosohed C+137: The concentration of Sisohed Cs in normal times (August and October 2012) multiplied by the river flow rate SS: The radioactive Cs concentration in Sisomeder Simples multiplied by the SS flow rate, which was obtained based on contiguous data from a turbidity meter and the river flow rate Largencies materians: a strand Simples multiplied by the SS flow rate, which was obtained based on contiguous data from a turbidity meter and the river flow rate Largencies atom data strand Simples multiplied by the SS flow rate, which was obtained based on contiguous data from a turbidity meter and the river flow rate Largencies atom data strand strandition and the upstream of the Udagava Werk: CT3 dreived form SS 5: The data indicated when human grading of the second int of Cs-137 deposited on soil and the su

4.4 Long-term Effects

Radioactive materials that adhered to tree leaves and branches immediately after the accident have transferred to the mulch layer and soil on the forest floor over time. At present, approx. 80% is retained in the soil surface layer and is strongly fixed in mineral soil.

Surveys conducted so far revealed that the annual outflow rate of Cs-137 from forest soil is around 0.02% to 0.3% of the total amount of Cs-137 deposited on nearby watershed soil.

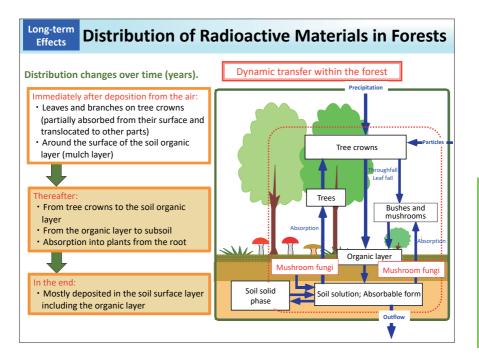
\* Prepared based on the material for the 16th meeting of the Environment Recovery Committee



Nuclear tests in the atmosphere were frequently conducted from late 1950s to early 1960s, causing a large amount of radioactive fallout across the globe. Radioactive cesium and radioactive strontium, etc. detected before March 11, 2011, are considered to be part of such fallout (p.74 of Vol. 1, "Effects of Radioactive Fallout due to Atmospheric Nuclear Testing").

As a result of a soil survey conducted in Hokkaido in 2009, Cs-137 was detected as deep as 40 cm from the ground surface in plowed soil, such as paddy fields and upland fields, but it was found that in forests where soil is not plowed, Cs-137 was mostly located within 20 cm from the ground surface.

How deep radioactive cesium is adsorbed in soil depends on the property of soil, but it is known that Cs-137 tends to remain in the surface layer also in Japan. (Related to p.172 of Vol. 1, "Distribution of Radioactive Cesium in Soil")



Distribution of radioactive materials in forests is considered to change significantly over years.

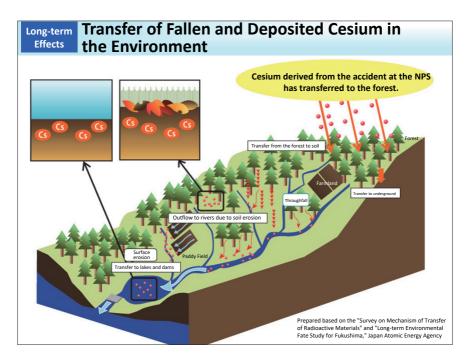
Radioactive cesium in the air adheres to leaves and branches, which eventually wither and turn into soil containing organic matter like muck soil. Some radioactive materials are absorbed from leaves or bark and transfer to new leaves or fruits within the plant, but they also turn into soil in the end.

Organic-rich soil lacks clay minerals that adsorb cesium and cesium tends to be absorbed into plants in such soil. For example, mushrooms often contain cesium at relatively high concentrations. This is partly due to the nature of the relevant mushrooms, but is also considered to have something to do with the fact that mushroom fungi grow well in organic-rich soil with less clay minerals.

Radioactive cesium in the organic layer gradually transfers into subsoil, and plants that take root deeper than the surface layer will come to absorb such cesium.

In this manner, radioactive cesium is fixed in the clayey soil in the process of circulating between plants and soil and is finally deposited in the surface layer of soil, as in the case of stable cesium.

As a result of the measurement of cesium in river water conducted by the Forestry and Forest Products Research Institute, cesium was not detected in most of the river water samples. Cesium was detected only in samples of turbid water collected on days with precipitation but the detected values were very small (P.23 of Vol. 2, "Readings of the Monitoring of Radioactive Cesium in Mountain Streams (2012)").

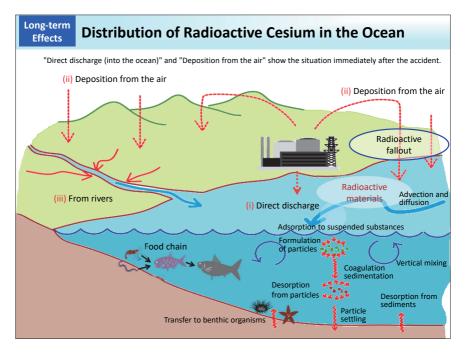


Distribution of radioactive cesium released into the environment due to the accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS has changed significantly over time. Cesium that adhered to tree bark, branches and leaves immediately after the accident transferred onto the forest soil due to leaf fall and precipitation, etc. At present, over 90% is found to be located within a depth of 5 cm from the ground surface. In the meantime, as the decrease in cesium at the ground surface is larger than the decrease due to physical attenuation, it is estimated that some cesium has transferred to underground.

Cesium has a property to be strongly adsorbed by specific clay minerals and is seldom dissolved in water. Furthermore, re-scattering into the air due to wind, etc. is hardly observed at present. Given these, outflow of cesium from forests to people's daily living areas is considered to be very minor.

The above figure illustrates the process that fallen and deposited cesium in the forest flows from the upstream to a downstream dam lake. The two enlarged pictures show the forest floor and the sediment at the bottom of the dam lake, both indicating that cesium is deposited in the surface layer of soil.

In a racing river, cesium is transported to the downstream while being adsorbed to soil particles, and in a gentle stream, cesium tends to be deposited onto river sediments. When there is a dam in the upstream, cesium is blocked at the dam lake and the amount that flows out to the downstream is smaller. Even when the water level of the dam lake becomes higher due to a typhoon or a heavy rain, the flow at the bottom sediments near the sluice is slow and deposited soil seldom raises up.



Distribution in the ocean of radioactive materials released due to the accident at Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS has changed significantly over time. There are three routes through which radioactive materials are transported to the ocean: (i) direct discharge of radioactive materials into the ocean from the NPS; (ii) fall onto the ocean of radioactive materials transported with wind; and (iii) transportation into the ocean of fallen radioactive materials via rivers or groundwater. However, in the case of cesium, which is strongly adsorbed in soil, it is hardly possible to imagine that it transfers together with groundwater and reaches the ocean.

Radioactive Cs concentrations in seawater increased significantly immediately after the accident but declined in one or two months as cesium was transported or diffused with the ocean current. Radioactive Cs concentrations in marine organisms, which have much to do with radioactive Cs concentrations in seawater, also declined in tandem with the decline in radioactive Cs concentrations in seawater. Additionally, transfer of radioactive cesium, part of which was deposited on the sea bottom, to bottom fish was a worry, but the survey results show declines in radioactive Cs concentrations in flatfish, Pacific cod, and other bottom fish including those caught off Fukushima Prefecture. It has been made clear that radioactive cesium has rarely transferred from sea-bottom soil to marine organisms (Source: "Report on Inspection of Radioactive Materials in Fishery Products" (2015), Fisheries Agency).

(Related to p.78 of Vol. 2, "Trends of Radioactive Cesium Concentrations by Fish Species (2/2)")

Long-term **Concentration Factors for Marine Organisms** Effects Concentration factor = (Radioactivity concentration in a marine organism) / (Radioactivity concentration in seawater) Concentration Types of organisms factor\* (cesium) 9 Squids and octopuses Phytoplankton 20 Zooplankton 40 Algae 50 Shrimps and crabs 50 Shellfish 60 Fish 100 Dolphin 300 The current radioactive cesium concentrations Sea lion 400 seawater are at the same level as that before the accident (0.001 - 0.01 Bg/L).

\* Concentration factors are recommended values in the following document by the IAEA.

Source: "Sediment Distribution Coefficients and Concentration Factors for Biota in the Marine Environment, 2004," International Atomic Energy Agency (IAEA)

The concentration factor is the ratio between the radioactivity concentration in a marine organism and the radioactivity concentration in seawater, assuming that the relevant marine organism is placed in seawater at a certain radioactivity concentration for a long period. This indicates the level of accumulation of radioactive materials in the relevant marine organism.

Comparing concentration factors of cesium, the concentration factor is higher for fish than plankton and is further higher for large mammals that eat fish.

Cesium also bioaccumulates, but is not continuously accumulated in organisms unlike mercury or cadmium. Instead, radioactive cesium concentrations in organisms are considered to decline in accordance with the decline in radioactive cesium concentrations in seawater.

Concentration factors indicated in the above figure are those recommended by the International Atomic Energy Agency (IAEA). At present, radioactive cesium concentrations in seawater have declined to almost the same level as that before the accident (0.001 - 0.01 Bq/L), except within the port near Tokyo Electric Power Company (TEPCO)'s Fukushima Daiichi NPS.