

B-16.7 Studies on Emission Control of CH₄ and N₂O from Solid Waste Disposal Facilities (Final Report)

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Abstract State and control methods of CH₄ and N₂O emission from solid waste disposal facilities were studied by means of *in situ* observation, laboratory experiments, published data and mathematical analysis. Firstly, the importance of the complete combustion in mechanical batch combustion type incineration facilities for the CH₄ emission control is confirmed by a *in situ* observation. Secondly, in night soil treatment plants with high loading dinitrification processes, it is suggested that some operation manners, such as the quality of substrate and duration of aerobic/anaerobic condition in reactor(s), would mainly control N₂O emission. Thirdly, at MSW landfill sites, the control of air filled porosity of cover soil is important for the CH₄ emission control by means of regulating the CH₄ consumption of soil.

1. Introduction

In the field of solid waste disposal, most CH₄ is emitted from MSW landfill sites by microbial digestion of organic wastes and N₂O is mainly emitted from incineration plants and night soil treatment plants by combustion and dinitrification, respectively. In this study, we attempt to extract some factors regulating the GHG emission by means of *in situ* observation, laboratory experiments, published data and mathematical analysis, and to establish control measures for the GHG emissions at facilities.

2. CH₄ emission control measures for MSW incineration plants

We observed the CH₄ emission from a mechanical batch combustion type incineration plant. Exhaust gas from the incinerator was piped to GC-FID via a automatic sampler with a syringe, and CH₄ concentration in composite gas samples (during 25 min) were continuously measured. **Fig.1** shows the changes of CH₄ concentration in exhaust gas from plants over 2 days. On 1st day, when waste load to the incinerator was low and combustion was completely shut down at the end of operation (i.e. the "burn-out" operation), CH₄ emission factor was relatively low (374 g-CH₄/ton-waste). On the other hand, CH₄ emission factor on 2nd day, when load was higher than that of 1st day and a flame was remaining in the residual

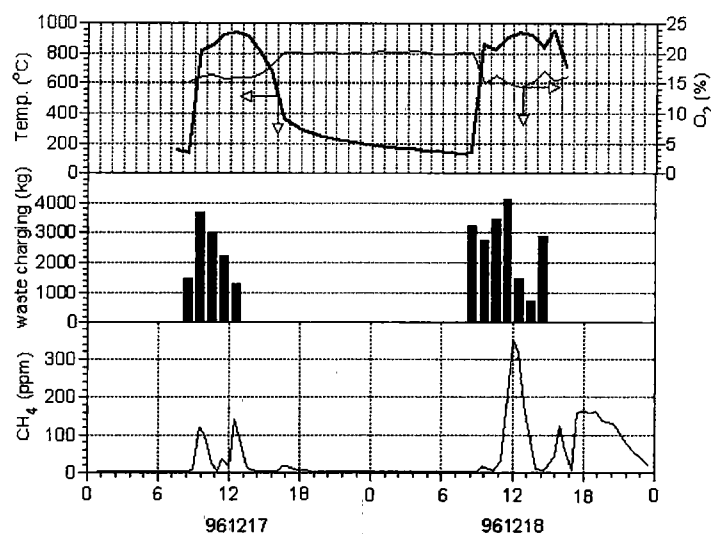


Figure 1 Changes of CH₄ concentration in exhaust gas from mechanical batch combustion type incineration plant

waste in incinerator after the end of operation (i.e. the “stock fire” operation which is widely used in this type of incinerators), was relatively high (697 g-CH₄/ton-waste). It confirmed that the need for measures to complete combustion in incineration facilities as possible for the CH₄ emission control.

3. N₂O emission control measures for night soil treatment plants

(1) Potential N₂O emission from night soil treatment plants

Analyzing information in literature¹⁾²⁾, we estimated the potential N₂O emission from night soil treatment plants in Japan. 25.6 % of total nitrogen (T-N) in night soil and gray water was charged into night soil treatment plants. Assuming the conversion factor of T-N to N₂O as 0.12-14%, potential N₂O emission from plants with high loading dinitrification processes was ca. 3% of total potential N₂O emission (0.014-1.6 Gg-N₂O/year) originated from night soil and gray water.

(2) N₂O conversion ratio at night soil treatment plants

We collected exhaust gas samples from selected 15 of night soil treatment plants with high loading dinitrification processes in Japan. Measurement of N₂O in gas and T-N in charged materials was used to estimate the conversion ratio of charged T-N to emitted N₂O. The ratio was ranged from 7.6% to 72%. Fig.2 shows that the N₂O conversion ratio tended to decrease with increase in COD/T-N of charged waste. It suggests that N₂O production from nitrification/dinitrification processes in plants would attribute to the deficient of the electron supplier, in other words, the incompleteness of

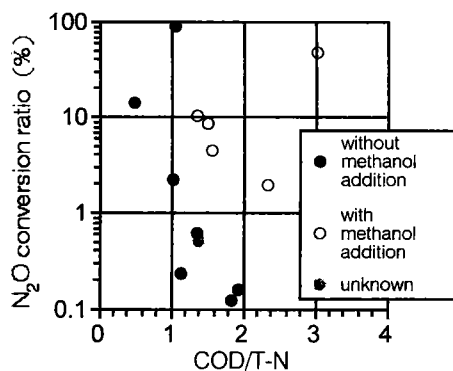


Figure 2 Relationship between N₂O conversion ratio and COD/T-N

dinitrification process³⁾.

(3) Continuous observation of N₂O emission from a night soil treatment plant with the high loading dinitrification process

In order to observe the diurnal variation of N₂O emitted from a night soil treatment plant with the high loading dinitrification process and the UF membrane separator, exhaust gas from the single type biological reactor was piped and introduced to IR N₂O analyzer via moisture traps. The mean N₂O concentration in exhaust gas was 3 ppm (0-36.5 ppm) during observation periods. Fig.3 shows the 2 patterns of N₂O fluctuation; sharp

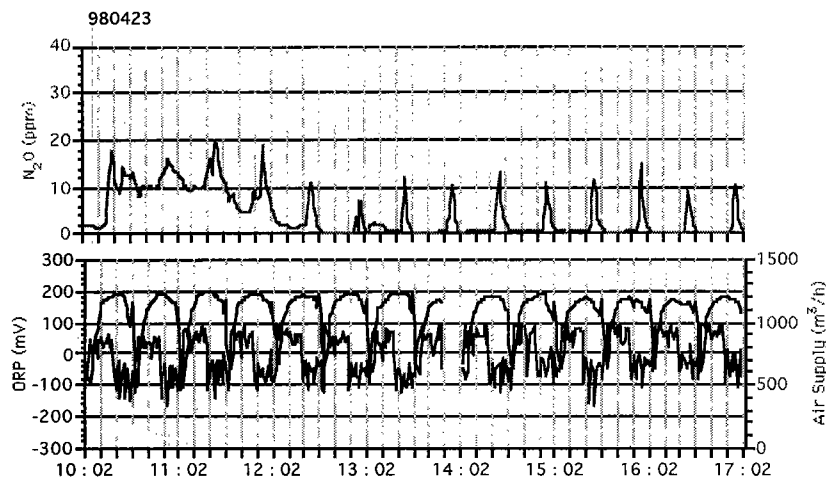


Figure 3 Changes of N₂O from exhaust gas from a night soil treatment plant with the high loading dinitrification process and the UF membrane separator

peaks of 10-40 ppm at intervals of 30 min and a broad peak of 10-20 ppm at 10:00 to 13:00. Former peaks appeared with decreasing step of air supplying to the reactor. While, duration of latter peak was consistent with the haulage period of new night soil and sludge from Joukasou. These trends are suggested that operation of the reactor and quality of substrate would be important in N₂O control measure in this type of plants.

4. CH₄ emission control measures for MSW landfill sites

(1) CH₄ emission from MSW landfill site in Japan, Korea and Indonesia

Field observation were performed at a MSW landfill site in Japan, Korea and Indonesia. CH₄ flux through the ground surface and from the gas suction pipe was estimated by the closed chamber and forced chamber method, respectively. Soil sample was collected at same point as surface flux measurement and biomass and activity of methane oxidizing bacteria (methanotrophs) in the cover soil were also estimated. For example, surface CH₄ flux in a Japanese landfill site is shown in Fig.4. These spatial and temporal distribution of CH₄ flux were also observed in Korean and Indonesian site, and we consider that it would depend on deposition period of refuse and physical status of cover soil, especially the air permeability of soil. Methane ratio ($\text{CH}_4/\{\text{CH}_4+\text{CO}_2\}$) of surface flux was mostly lower than that of gas from suction pipes, which is considered to represent original landfill gas from the garbage layer. Since methanotrophs were omnipresence through landfill cover soil and have relative high capacity of CH₄ oxidation (Fig.5), methane oxidation in cover soil would also regulate the landfill gas flux.

(2) Development of Mobile Q-MS methane analyzer

Since surface CH₄ flux is substantially distributed in landfill sites, number of points for

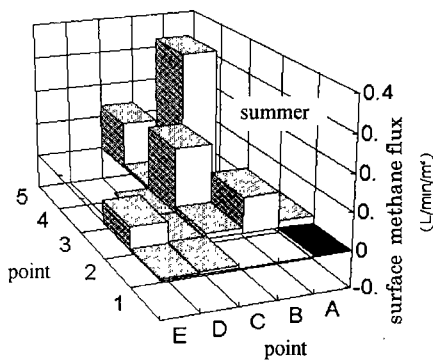


Figure 4 Distribution of surface CH_4 flux in a Japanese landfill site

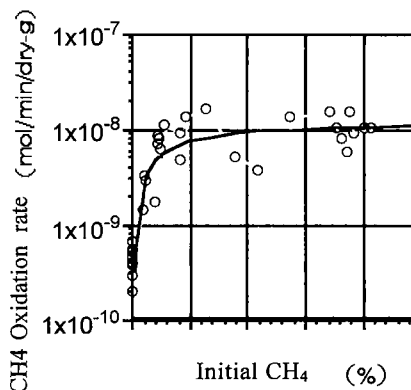


Figure 5 CH_4 oxidation rate of cover soil

observation are required to estimate the total CH_4 emission as possible. We, therefore, mobile CH_4 analyzer was developed using Q-MS detector. This analyzer has the detection limit of 100 ppm- CH_4 and principally not need the carrier gas.

(3) Numerical analysis of methane oxidation and migration in landfill cover soil

We theoretically estimated the surface CH_4 flux by a primary mathematical model including landfill gas migration and CH_4 consumption in cover soil. According to Tanaka and Kohyama⁴⁾, a model included the Darcy's law for the flow of soil gas in porous media under the pressure gradient between the refuse layer and atmosphere and Stephan-Maxwell equations for the bulk diffusion of multicomponent gas in porous media. CH_4 oxidation rate, which is formulated using Michaelis-Menten type equation depending on growth substrates such as CH_4 and O_2 in cover soil was also included in the model as the reaction term. Setting the several parameter and boundary conditions from our experiments, literature and some assumptions to the model, we calculated the CH_4 flux in the soil column of 1m at intervals of 0.1m using the finite difference method. **Fig.6** shows ratio of CH_4 emission at surface (top boundary) without to with CH_4 oxidation in cover soil. Calculations indicated that air filled porosity and total pressure of landfill gas should greatly affect to CH_4 oxidation in soil. Thus, the air filled porosity, i.e. air permeability of soil is considered as one of major and applicable factor in emission control measures. **Fig.7** shows the ratio of CH_4 oxidation when the pressure difference between atmosphere and garbage layer is 0 and landfill gas not passing through soil will directly release to atmosphere via short-cut passways such as gas suction

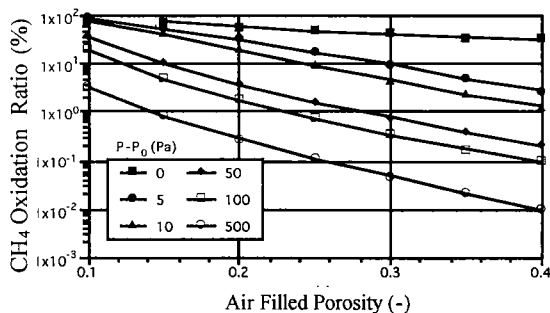


Figure 6 CH_4 oxidation ratio under various porosity and pressure conditions

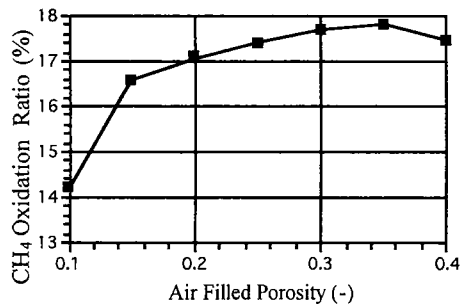


Figure 7 CH_4 oxidation ratio at pressure difference = 0 with assumption that landfill gas not passing through soil will directly release to atmosphere

pipes and cracks. We can find the optimal air filled porosity to maximize CH₄ oxidation in figure. This illustration would be one concept for design and application of cover soil with regards of CH₄ suppression.

5. Conclusion

State and control methods of CH₄ and N₂O emission from solid waste disposal facilities were studied by means of *in situ* observation, laboratory experiments, published data and mathematical analysis. Main findings are presented in follows.

- (1) In a *in situ* observation, it is confirmed that the complete combustion in mechanical batch combustion type incineration facilities need for the CH₄ emission control.
- (2) In night soil treatment plants with high loading dinitrification processes, it is suggested that some operation manners, such as the quality of substrate and duration of aerobic/anaerobic condition in reactor(s), would mainly control N₂O emission.
- (3) At MSW landfill sites, the control of air filled porosity of cover soil is important for the CH₄ emission control by means of regulating the CH₄ consumption of soil.

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Reference

- 1) MHW (1997): Waste Management in Japan 1996, Japan Waste Management Association
- 2) I. Watanabe, Y. Matsuzawa, M. Osako, M. Yamada and M. Tanaka (1997) : Emission of Nitrous Oxide from Processes of Night Soil Treatment,, Proceedings of 7th International Workshop on Nitrous Oxide Emissions, Cologne, Germany,, April 1997
- 3) H. Zeng, K. Hanaki and T. Matsuo (1994) : Production of nitrous oxide gas during nitrification of wastewater, Preprint Book, 1, IAWQ Biennial International Conference, Budapest, 117-125
- 4) Tanaka, N. and K. Koyama (1991): Gas and Gas Components Migration in Sanitary Landfill Sites, J. Jap. Soc. Waste Management Experts, 2: 46-58 (in Japanese)